Research Article

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Safe and symptomatic medicinal use of surface-functionalized Mn₃O₄ nanoparticles for hyperbilirubinemia treatment in mice



Aim: Testing the potential of citrate-capped Mn₃O₄ nanoparticles (NPs) as a therapeutic agent for alternative rapid treatment of hyperbilirubinemia through direct removal of bilirubin (BR) from blood in mice. **Materials & methods:** NPs were synthesized and the mechanism of BR degradation in presence and absence of biological macromolecules were characterized *in vitro*. To test the *in vivo* BR degradation ability of NPs, CCl₄-intoxicated mice were intraperitoneally injected with NPs. **Results:** We demonstrated ultrahigh efficacy of the NPs in symptomatic treatment of hyperbilirubinemia for rapid reduction of BR in mice compared with conventional medicine silymarin without any toxicological implications. **Conclusion:** These findings may pave the way for practical clinical use of the NPs as safe medication of hyperbilirubinemia in human subjects.

Keywords: hyperbilirubinemia • mice model • Mn-based functionalized nanoparticles • nanomedicine • nanotherapy of jaundice

The term hyperbilirubinemia is defined as increased bilirubin (BR) level (>1.3 mg/dl in human) in blood. Hyperbilirubinemia is caused when there is an imbalance between production of BR (resulting from hemolysis, sepsis, blood extravasation or polycythemia) and decrease in BR excretion due to inadequate hepatic conjugation and increased enterohepatic reabsorption (resulting from pyloric stenosis, delayed bacterial gut colonization, GI tract immobility or obstruction) [1-3]. Although at micromolar concentrations BR acts as an antioxidant for scavenging peroxyl radicals in blood [4], yet it can be toxic and harmful to cells at higher concentration [5,6]. Elevated level of BR and its oxidative products in human blood causes various diseases including neonatal jaundice (>10 mg/dl), Gilbert syndrome (with BR level reaching >6 mg/dl), Crigler–Najjar type I disease (>30 mg/dl) [7] and BR-induced neurologic dysfunction [8]. Severe neurotoxicity in case of neonates (Kernicterus) and damage in white matter of adult brain are also the consequences

of higher BR level [9,10]. In case of hepatitis E, infection in pregnant women, associated hyperbilirubinemia itself is found to increase the risk of preterm delivery [11]. Treatment options for hyperbilirubinemia include phototherapy [12], hemoperfusion, hemodialysis and exchange blood transfusion [13]. Although phototherapy is widely used and a best choice for the treatment of hyperbilirubinemia in neonates, its efficacy is noted to diminish as children with Crigler-Najjar syndrome advance in age and in particular during adolescence [12,14,15]. Hemoperfusion and exchange blood transfusion have significant morbidity and even mortality [1]. Apart from the mentioned therapeutic procedures various organic, inorganic and phytochemicals have been used for reduction of elevated BR level in blood [16-19]. All these therapeutic drugs function in hepatoprotective way and lack the ability to degrade BR directly [2]. So, none of these became very fruitful in effective and quick reduction of serum BR level. In the above context, a safe, symptomatic and effective therapeutic stratNabarun Polley¹, Srimoyee Saha², Aniruddha Adhikari¹, Somtirtha Banerjee², Soumendra Darbar³, Sukhen Das² & Samir Kumar Pal^{*,1} 'Department of Chemical, Biological

& Macromolecular Sciences, S N Bose National Centre for Basic Sciences, Block JD, Sector III, Salt Lake, Kolkata 700 098, India

²Department of Physics, Jadavpur University, 188, Raja Subodh Chandra Mullick Road, Jadavpur, Kolkata 700032, India

³Research & Development Division, Dey's Medical Stores (Mfg.) Ltd, 62, Bondel Road, Ballygunge, Kolkata 700019, India *Author for correspondence: skpal@bose.res.in



egy is extremely needed, and is the principal motive of the present work.

Manganese (Mn), one of the safest materials is a natural constituent, and often a cofactor for enzymes and receptors [20]. In a series of earlier studies from this group, we have established that the charge state of Mn in the citrate-capped Mn₃O₄ nanoparticles (C-Mn₃O₄ nanoparticles [NPs]) is involved in selective catalysis of BR in absence of any photoactivation [21,22]. The mixed valence state of Mn (+2, +3 and +4) along with the functional groups on the surface coordinating ligands of Mn₃O₄ NPs leads to the exceptional catalytic activity. The very fast and selective reduction of the BR level (both conjugated and unconjugated) in the blood specimens in *ex vivo* condition without any significant alteration of other essential blood parameters have also been reported.

In the present study, we have synthesized citratecapped Mn₃O₄ NPs by ligand-etching technique as described in earlier studies [21,23]. The structural characterization and the interaction with a model-blood protein of the NPs have been performed using high resolution transmission electron microscopy (TEM), steady state and time-resolved optical spectroscopy. Ultrahigh efficacy of generation of reactive oxygen species (ROS) and surface-mediated catalysis of the synthesized NPs, have also been established. We have also performed systematic in vitro and in vivo preclinical studies on the toxicity of the functionalized NPs. All the measurements confirm that the Mn-based nanomedicine is safe and biocompatible due to lack of any potential toxicity. In order to assess the effectiveness of the NPs as symptomatic nanomedicine in managing hyperbilirubinemia in mice, we have checked the BR concentration and found to become almost normal in 6 h after intraperitoneal injection of the NPs. Therefore, these results demonstrate that the functionalized NPs are strong candidate as effective drug for hyperbilirubinemia treatment, and superior to the commercially available drug, silymarin in terms of efficacy with a satisfactory biocompatibility.

Materials & methods

Materials

Manganese chloride, sodium citrate, NaOH, BR, human serum albumin (HSA), silymarin, heparin, acrylamide, bis-acrylamide, ammonium persulfate, isobutanol, tris base, bromophenol blue, dithiothreitol, tetramethylethylenediamine were obtained from Sigma-Aldrich (MO, USA). 2', 7'-dichlorofluorescin diacetate (DCFH-DA), Ethanol amine, HCl and glycerol were received from Merck (NJ, USA). All the chemicals used for this study were of analytical grade and were used without further purification.

Synthesis of citrate-functionalized Mn₃O₄ NPs

Synthesis of bulk Mn₂O₄ NPs was done following a reported ultrasonic-assisted approach for preparation of colloidal Mn₃O₄ NPs at normal temperature and pressure without any additional surfactants or templates [23]. For functionalizing the as-prepared Mn₂O₄ NPs by citrate ligand first, 0.5 M citrate (ligand) solution was prepared in Milli-Q (from Millipore) water. Then, the pH of the solutions at approximately seven was adjusted by drop wise addition of 1 (M) NaOH solution. In the ligand solution of pH approximately seven, as-prepared Mn₃O₄ NPs (~100 mg of powder Mn₂O₄ NPs in 5 ml ligand solution) were added and extensively mixed for 12 h in a cyclomixer. Finally, using a syringe filter of 0.22 µm diameter the nonfunctionalized bigger-sized NPs were filtered out. The resulting filtrated solutions were used for our experiments, both in vivo and in vitro studies without further dilution [21].

Microscopic & spectroscopic studies

For the characterization of the NPs, TEM samples were prepared by dropping sample stock solutions onto a 300-mesh carbon-coated copper grid and dried overnight in air. Particle sizes were determined from micrographs recorded using an FEI TecnaiTF-20 fieldemission high-resolution transmission electron microscope operating at 200 kV. Optical absorbance spectra of the solutions were recorded using a quartz cuvette of 1 cm path length in Shimadzu Model UV-2600 spectrophotometer. For the fluorescence imaging study of liquid NPs, Olympus BX51 fluorescence microscope was used. Jobin Yvon Model Fluoromax-3 Fluorimeter was used to study the characteristic fluorescence excitation and PL spectra of C-Mn₃O₄ NP solutions. Far UV circular dichroism measurements were performed on a JASCO815 spectrometer (from JASCO International Co. Ltd, Tokyo, Japan) by using a 1 cm path-length quartz cell. Fluorescence transients were measured using the 16-channel time-correlated singlephoton counting (TCSPC) setup, assembled in the lab with all the components (16-channel PMT module PML-16-1-C, Simple Tau-130EM with SPC-130EM & DCC-100 cards, Express Card 54 and SPCM64 software preinstalled in Lenovo ThinkPad laptop-PC) from Becker & Hickl (Berlin, Germany).

For 283-nm laser excitation, we used a femtosecond coupled TCSPC setup [24]. Briefly, the samples were excited by the third harmonic laser beam (283 nm) of the 850 nm (0.5 nJ per pulse) by using a mode-locked Ti-sapphire laser with an 80 MHz repetition rate (Tsunami, Spectra Physics), pumped by a 10W Millennia (Spectra Physics) followed by a pulse-picker (rate 8 MHz), and a third harmonic generator (Spectra-Physics, model 3980). The third harmonic beam was used for excitation of the sample inside the TCSPC instrument (instrument response function (IRF = 180 ps) and the second harmonic beam was collected for the start pulse. Fluorescence transients were fitted by a nonlinear least square fitting procedure to a function

$$x(t) = \int_0^t E(t')R(t-t')dt'$$

comprising the convolution of the IRF (E [t]) with a sum of exponentials

$$R(t) = A + \sum_{i=1}^{N} B_i e^{\frac{\tau}{\tau}}$$

characteristic lifetimes (τ_i) , and a background (A). Relative concentration in a multiexponential decay was finally expressed as:

$$C_n = \left(\frac{B_n}{\sum_{i=1}^{N} B_i}\right) \times 100$$

The quality of the curve fitting is evaluated by a reduced chi-square and residual data.

Other measurements

In order to investigate the efficacy of the NPs for the generation of ROS, DCFH (dichlorofluorescein) was prepared from DCFH-DA at room temperature by mixing 0.5 ml of 1.0 mM DCFH-DA in methanol with 2.0 ml of 0.01N NaOH [25-27]. The solution was then neutralized with 10 ml of 25 mM NaH₂PO₄, at pH 7.4 and was kept on ice in the dark until use. Native gel electrophoresis studies intend to discriminate between HSA and HSA-NP (after 12 and 24 h of incubation) was performed using native PAGE. The gel was prepared according to standard procedures [11], without the addition of SDS or β -mercaptoethanol. Gradient in the gel was nonlinear. Electrophoresis was conducted in denaturing condition for about 3 h in a vertical apparatus (Aplex 305II) at room temperature, at 120V. After completion, protein bands were stained using Coomassie blue.

Animals

Swiss albino mice of either sex (5–8-weeks old, weighing 28 ± 4 gm) were procured from a Committee for the Purpose of Control and Supervision of Experiments on Animals (CPCSEA) approved animal house (registration number: 50/PO/99/CPCSEA) and housed in standard, clean polypropylene cages in a temperature controlled animal room (temperature $22 \pm 3^{\circ}$ C; relative humidity $45-60 \pm 1\%$; 12 h light/ dark cycle). Water and standard laboratory pellet diet for mice (Hindustan Lever, Kolkata) were available *ad libitum*. All mice were allowed to acclimatize for

1 week prior to experimentation. The animals were maintained according to the guidelines recommended by CPCSEA, New Delhi, India and approved by the Institutional Animal Ethics Committee (Approval number-JU-Dey's/IAEC/09/14, dated 31.01.2014).

Acute toxicity study

Single-dose intraperitoneal toxicity study was conducted to determine the possible acute toxicity of surface-modified Mn_3O_4 NPs following the general principles of the OECD guideline 423 with some adjustments [28]. Twelve female mice were divided in four groups: one control group (received 0.2 ml Milli-Q water) and three experimental groups (received either 600, 2000 or 5000 mg/kg bodyweight (BW) of NP dissolved in 0.2 ml Milli-Q water). All the animals were kept in fasting condition overnight prior to the injection. Behavior, mortality and BW were monitored daily for a period of 14 days.

Treatment protocol: Phase I

Forty eight mice of either sex were randomly divided into six groups of eight mice each. Group I served as the vehicle control and was given olive oil daily (0.5 ml/kg BW) for a period of 2 weeks. Group III served as NP control and left untreated for first 2 weeks, then received citrate-capped Mn_3O_4 NP (500 mg/kg BW dissolved in 0.2ml Milli-Q water) daily for last 7 days. For inducing hyperbilirubinemia and hepatotoxicity (in vivo), animals of groups II, IV, V and VI were administered with carbon tetrachloride solution (50% CCl₄ in olive oil) 1 ml/kg BW daily for a period of 2 weeks. CCl₄ is a well-known hepatotoxic agent frequently used to study hepatoprotective activity of new drugs in *in vivo* experimental model. CCl₄ administration induces critical liver damage in mice which in turn simulates a condition of acute hepatitis showing similar symptoms as humans [29-31]. CCl₄ activated by liver enzyme cytochrome P450 forms the trichloromethyl free radical (CCl,•) that damages hepatic cells [32]. These radicals bind covalently to sulfhydryl groups of glutathione and protein thiols in cells to initiate a chain of events leading to membrane-lipid peroxidation and cell necrosis [33-35]. After CCl₄ intoxication, group III served as the CCl₄ control and was left untreated. Group IV was administered with citrate-capped Mn₂O₄ NP (500 mg/kg BW dissolved in 0.2 ml Milli-Q water) for 7 days. Group V served as citrate control and was treated with citrate (100 µl/mice; 1:1 solution in water). Group VI served as the positive control and was administered silymarin (100 mg/kg BW) [36,37] daily for a period of 1 week. All the treatments were executed by intraperitoneal injection. At the end of the experiment, the animals were

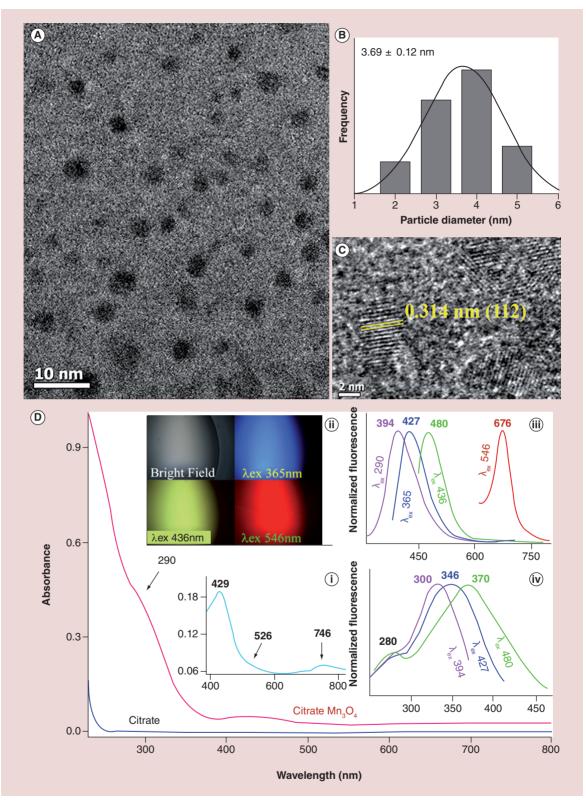


Figure 1. Characterization of nanoparticles. (A) The particle distribution of the nanoparticles (NPs) recorded under transmission electron microscopy. **(B)** The particle size distribution is shown with average size of 3.69 ± 0.12 nm and **(C)** high-resolution electron micrograph of the NPs with a clear interplaner distance of 0.314 nm for 112 planes in the Mn₃O₄ NPs is evident.

Figure 1. Characterization of nanoparticles (cont.). (D) The absorption spectra of citrate-capped NPs used as nanomedicine in the present study and the capping agent citrate are shown in panel. A magnified version of the absorption spectrum of the NPs in the rage of 400–800 nm is shown in the inset (i). Different absorption peaks due to d-d transitions (see text) are evident. Fluorescence microscopic view of the liquid solution of the NPs with different excitation is shown in the inset (ii). The corresponding emission and excitation spectra of the solution are shown in the insets (iii& iv), respectively.

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kept in fasting condition overnight and sacrificed by cervical dislocation.

Treatment protocol: Phase II

For studying the efficiency of BR degradation by citrate-capped Mn_3O_4 NP over standard drug silymarin, 48 mice were divided into four groups (n = 12 per group). Group I served as vehicle control and received olive oil daily (0.5 ml/kg BW) for a period of 3 weeks. Other four groups were administered with CCl_4 solution (50% CCl_4 in olive oil) 1 ml/kg BW daily for a period of 3 weeks to induce hyperbilirubinemia. After induction, group II left untreated, group III was administered with citrate-capped Mn_3O_4 NP (500 mg/kg BW dissolved in 0.2 ml Milli-Q water) twice daily and group IV received silymarin (100 mg/kg BW) twice daily. Serum biochemical tests were performed from blood collected at 2, 6, 12 and 24 h.

Measurement of serum biochemical parameters

For biochemical studies, blood samples were collected just before sacrifice in sterile tubes (nonheparinized) from retro-orbital plexus and allowed to clot for 45 min. Serum was separated by centrifugation at 3000 rpm for 15 min. All serum samples were sterile, hemolysis-free, and were kept at 4°C before determination of the biochemical parameters. Serum BR concentration (total and direct) was measured using commercially available test kits and results were expressed as mg/dl. Liver damage was assessed by the estimation of serum activities of alanine aminotransferase (ALT, EC 2.6.1.2), aspartate aminotransferase (AST, EC 2.6.1.1), and alkaline phosphatase (ALP, EC 3.1.3.1), y-glutamyltranferase (GGT, 2.3.2.2) using commercially available test kits. The results were expressed as units/liter (IU/l). Total protein concentration was estimated and expressed as gm/dL. All the kits for measurement of serum biochemical parameters were purchased from Autospan Liquid Gold, Span Diagnostics Ltd (Surat, India). All hematological tests were performed spectrophotometrically following the protocols described by the corresponding manufacturers.

Hematological study

For hematological studies, the blood was collected in heparinized tubes. Blood-cell count was done using blood smears in Sysmax-K1000 Cell Counter. Parameters studied were hemoglobin, total red blood cell, reticulocyte, hematocrit, mean corpuscular volume, mean corpuscular hemoglobin, mean corpuscular hemoglobin concentration, platelets, total white blood cell.

Histopathological examination

The liver was excised immediately after collection of blood, washed with ice-cold phosphate buffer and dried with tissue paper. It was weighed and fixed in neutral formalin solution (10%), dehydrated in graduated ethanol (50–100%), cleared in xylene and embedded in paraffin. Sections 4–5 μ m thick were prepared using microtome and then stained with hematoxylin and eosin (H&E) dye and examined for histopathological changes under the microscope.

Statistical analysis

All quantitative data are expressed as mean \pm standard deviation. Comparison of different parameters between the groups was performed using one-way analysis of variance followed by Tukey's multiple comparison test using a computer program GraphPad Prism (version 5.00 for Windows), GraphPad Software, California, USA [38]. p < 0.05 was considered significant.

Results & discussion

The spherical Mn₃O₄ NPs were imaged by TEM (Figure 1A) and statistical evaluation found a size distribution of about 3.69 ± 0.12 nm (Figure 1B). Figure 1C is the high resolution TEM image of an isolated Mn₃O₄ NP. The observed interfringe distance of 0.314 nm is consistent with Mn₂O₄ interplanar distance of 112 plane revealing crystalline nature of the NP [21]. The absorbance spectra of the NPs and the cappingligand citrate are represented in Figure 1D. A magnified view of the absorption spectrum of the NPs in the range of 400-800 nm is shown in the inset Figure 1Di. The absorbance maxima at around 290 and 429 526, 746 nm correspond to the high-energy ligandto-metal charge transfer transition involving citrate-Mn⁴⁺ interaction and Jahn–Teller (J–T) distorted d–d transitions centered over Mn³⁺ ions respectively [21,22]. The NPs are also found to be fluorescent under various excitation wavelengths as reported earlier [21]. A fluorescence microscopic image containing bright field and fluorescence from the NPs solution under UV (365 nm), blue (436 nm) and green (546 nm) excita-

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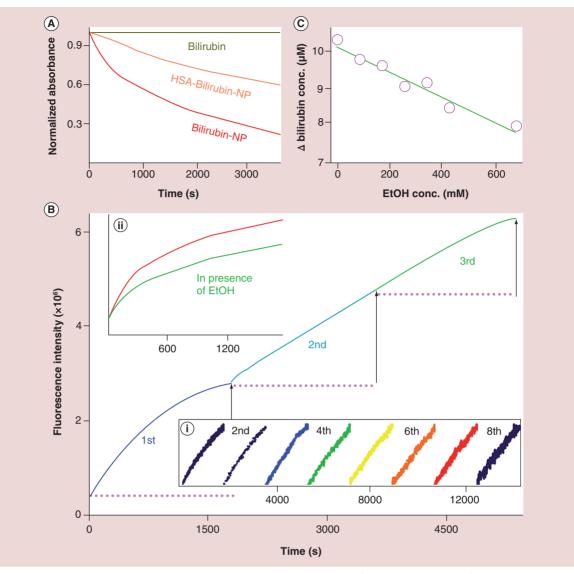


Figure 2. Bilirubin degradation by nanoparticles. Degradation kinetics of bilirubin in solution (*in vitro*) in presence of the nanoparticles (NPs) are shown in **(A)**. Effects of encapsulation of bilirubin in human serum albumin protein and the controlled kinetics of bilirubin in absence of NPs in our experimental condition are also shown in the panel. The efficacy of generation of reactive oxygen species, the key for the degradation of bilirubin for a number of cycles are shown in **(B)**. Reusability of the NPs (inset **[i]**) over eight cycles and reduction of the efficacy in presence of reactive oxygen species quencher ethanol (inset **[ii]**) are also shown. A linear dependency concentration of bilirubin degradation in 1 h on the concentration of ethanol is shown in **(C)**. conc.: Concentration.

tion is shown in the inset Figure 1Dii. The corresponding emission spectra of the NPs solution are shown in Figure 1Diii, revealing distinct emission peaks for different excitation wavelengths. Different peaks in the excitation spectra (Figure 1Div) form the solution for different detection wavelengths, confirm the multiple excited state in the emission from the NPs as detailed in earlier literature [21].

In order to investigate the efficacy of the NPs in BR degradation *in vitro*, we have monitored BR decomposition at pH = 7.4 in presence and absence of the

serum protein HSA (Figure 2A). We have maintained the concentration of the NPs similar to that used in the *in vivo* experiment (from same stock solution). The characteristic absorbance of BR at 450 nm has been chosen for monitoring the kinetic study using UV-visible spectroscopy. Figure 2A shows the relative concentration (C_t/C_0) of BR plotted against reaction time. In absence of HSA, decomposition of BR after 60 min by the NPs can be observed to be around 80%, whereas in presence of HSA, the decomposition reduced to 40%. In contrast to the catalytic decomposition of BR, only 2-5% of BR degradation is observed without the NPs. These results are self-explanatory, since the water soluble BR gets either higher chance to interact with NPs than in the albumin protein or more accessible to the NPs-generated radical in the aqueous solution [21]. Here, we have found that the NPs are efficient generator of ROS in aqueous solution. Enhanced emission of an ROS indicator DCFH is used for the monitoring of oxidative stress in the solution. The emission maxima of DCFH (520 nm) upon excitation at λ =488 nm was plotted against time to generate the fluorescence kinetic plot (Figure 2B) for repetitive addition of DCFH in the solution. The NPs were capable of producing ROS for at least eight cycles as highlighted in Figure 2Bi. The reduction in oxidative stress generation has also been observed in presence of well-known ROS quencher ethanol [21] in the medium (Figure 2Bii). Figure 2C clearly indicates that the generated ROS was inefficient to degrade the BR with increased level of ethanol concentration.

The observed reduction efficacy of the NPs in presence of HSA (Figure 2A) could be rationalized in the following two ways. First, the BR may interact directly with the NPs and undergoes surface-mediated catalysis (Langmuir-Hinshelwood type) and encapsulation of the BR in the presence of HSA hinders the interaction leading to less degradation of BR. Second, the NPs could be encapsulated by the protein HSA for the less interaction with the BR in the aqueous solution. We have ruled out the first possibility by an attempt to study the rate of BR degradation kinetics with the concentration of the NPs in the solution. The rate of the kinetics was found to be independent of the NPs concentration, which should not be the case of surfacemediated catalysis. In order to address the second possibility, we have investigated the interaction of the NPs with the protein HSA as shown in Figure 3. The far-UV circular dichroism spectra of HSA, exhibited two negative bands in the UV region at 208 and 220 nm (Figure 3A), characteristic of an α -helical structure of protein [39]. The percentage of helix content of HSA both in presence and absence of NPs is found to be similar revealing insignificant conformational alteration in α-helical structure of HSA in presence of the NPs. The molecular mass of the protein HSA in presence of the NPs is also found to be intact as shown in native gel electrophoresis study (Figure 3B). A direct evidence of the nature of interaction of the NPs with the protein is clear from picosecond resolved fluorescence transient of the single tryptophan Trp-214 (excitation 283 nm) in absence and presence of the NPs as shown in Figure 3C (average lifetime of 1.95 ns). In case of proximity of the NPs to Trp-214, a quenching of the fluorescence is unavoidable due to Förster

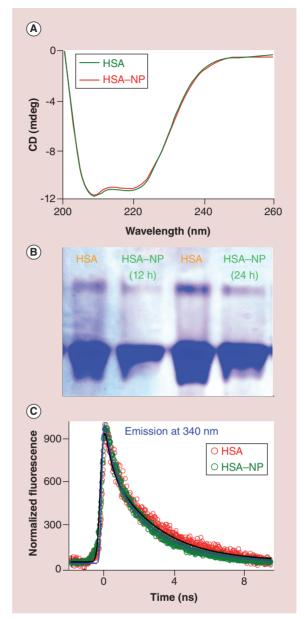


Figure 3. Human serum albumin-nanoparticle interaction. (A) Circular dichroism spectra of HSA without and with the NPs (see text). An insignificant effect of the NPs in the secondary structure of the serum protein is evident. (B) The overall structural integrity of the protein upon interaction with the NPs with various incubation time is shown from the mobility of the protein in a native gel. (C) Picosecond resolved transient of the single tryptophan (Trp-214) of the serum protein without and with the NPs. In case of direct interaction of the NPs with the protein, guenching (due to Förster resonance energy transfer) of the transient in presence of the NPs is unavoidable because of the strong spectral overlap of the tryptophan emission and absorption of the NPs. However, insignificant quenching of Trp-214 in presence of NPs rules out the possibility of direct interaction (see text).

HSA: Human serum albumin; NP: Nanoparticle.

resonance energy transfer (FRET) [40] as a result of strong spectral overlap of the Trp-214 emission with the absorption of the NPs (Figure 1D). Insignificant change in the Trp-214 emission transient in presence of the NPs, clearly rules out proximal interaction of the protein with the NPs.

For assessing the maximal-tolerated dose of surfacefunctionalized Mn₂O₄ NP, we performed single-dose acute toxicity study as per OECD guideline. Intraperitoneal injection of Mn₃O₄ NP did not cause any mortality throughout the experimental period for all three dose groups. Behavioral and physical symptoms of acute toxicity such as decreased activity or decreased uptake of food and water were not also been observed during the study period. As no clinical signs of toxicity were observed upto 5000 mg/kg BW, hence 1/10th of the maximum dose administered (i.e., 500 mg/kg BW) was selected for the present study.

In the phase I study, the total serum BR (Tsb) level of CCl₄ treated group (group II) increased to 0.89 ± 0.07 mg/dl which is far higher than the control group (group I; $0.32 \pm 0.04 \text{ mg/dl}$), indicative of hyperbilirubinemia. This group (group II) was left untreated for a week to see the auto recovery. The serum biochemical tests were done after treatment of seven days and represented in Table 1. The results show that treatment with the NPs significantly decreased BR level (total and direct) even lower than the sham control (Figure 4A & B). But for group II, we have seen very mild decrease in serum BR concentration even after removal of the hepatoxin, indicating slow recovery process. However, in case of direct BR level, the reduction is not significantly different from the clinical standard silymarin (Table 1). The NPs directly interacts with the BR and degrade the same. In the case of direct BR (soluble in aqueous environments and complex with

glucuronic acid), hindrance of the interaction with NPs could be a possible reason of lower efficacy of the NP in reducing direct BR.

For assessing the efficiency of BR degradation in vivo by NPs over standard drug silymarin, we performed the second phase of experiment. Here, BR concentration was monitored for a period of 24 h after intoxication. Three weeks CCl₄ intoxication increased the total serum BR level upto 1.28 ± 0.14 mg/dl. Single dose of $Mn_{3}O_{4}$ NP decreases Tsb to 0.40 ± 0.07 mg/dl within 2 h and back to normal level ($0.32 \pm$ 0.05 mg/dl) within 6 h compared with the silymarin treated group 1.22 ± 0.16 mg/dl (2 h) and $1.10 \pm$ 0.14 mg/dl (6 h), respectively. The results are represented in Figure 4C & D. Treatment with NPs decreased Tsb level almost 70% compared with that of silymarin (8%) in 2 h and restored normal Tsb concentration within 6 h. The results suggest the efficiency of citratefunctionalized Mn₃O₄ NP in targeted degradation of BR in a much faster rate compared with that of standard drug silymarin in in vivo experimental system.

For assessment of liver injury, serum activity of various hepatic lysosomal enzymes is used as diagnostic indicators [41]. Earlier studies have shown that CCl₄ administration increases serum levels of AST, ALT, ALP and GGT [16-19,42-45]. Normal liver contains high concentrations of these enzymes. When there is hepatocyte necrosis or membrane damage, these enzymes are released into the circulation, as indicated by elevated serum enzyme levels [46]. The hepatotoxicity of CCl₄ was confirmed in our study by a significant elevation of serum ALT (197.86%), AST (157.63%), ALP (106.77%) and GGT (105.36%) activities in the CCl₄-treated group as compared with the vehicle control which are summarized in Table 2. However, treatment with NPs at a dose of 500 mg/kg BW for 7 days

Group	Design of treatment	Total bilirubin (mg/dl)		Direct bilirubin (mg/dl)		
		After induction	After treatment	After induction	After treatment	
I	Sham control	0.33 ± 0.05	$0.32 \pm 0.04^{+}$	0.19 ± 0.008	$0.20 \pm 0.009^{\dagger,\$}$	
II	CCl ₄ control	0.89 ± 0.07	$0.58 \pm 0.04^{\pm,\$}$	0.39 ± 0.054	$0.33 \pm 0.044^{\pm,\$}$	
III	NP control	0.34 ± 0.04	$0.18 \pm 0.04^{\text{t,t,§}}$	0.21 ± 0.009	$0.11 \pm 0.019^{+,\pm}$	
IV	CCl ₄ + NPs	0.93 ± 0.06	$0.19 \pm 0.03^{+,+,\$}$	0.42 ± 0.045	$0.13 \pm 0.019^{+,\pm}$	
V	CCl_4 + citrate	0.88 ± 0.06	0.52 ± 0.05 ^{‡,§}	0.38 ± 0.036	$0.32 \pm 0.034^{\pm,\$}$	
VI	CCl₄ + silymarin	0.87 ± 0.07	$0.33 \pm 0.04^{+}$	0.38 ± 0.038	0.13 ± 0.019 ^{†,‡}	

One-way analysis of variance Tukey post hoc:

[†]p < 0.05 compared with CCl

 p^{*} < 0.05 compared with vehicle control.

[§]p < 0.05 compared with silymarin. Dosage: Olive oil: 0.5 ml/kg bodyweight; CCl, + olive oil sol.: 1 ml/kg bodyweight; NPs: 500 mg/kg bodyweight; silymarin: 100 mg/kg

bodyweight; Citrate (1:1 solution in water): 100 µl/mice

NP: Nanoparticle.

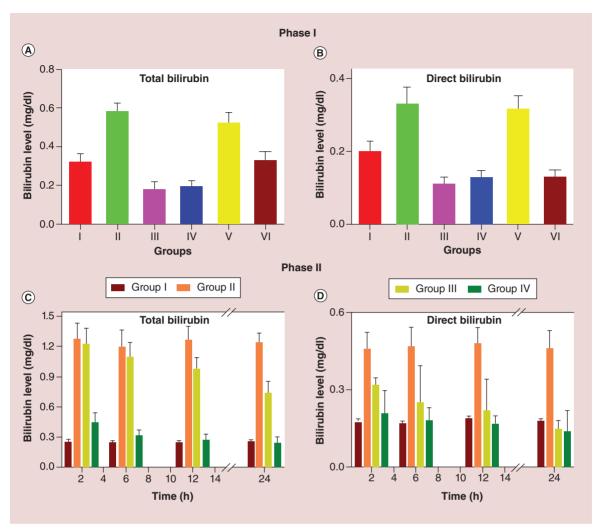


Figure 4. Effect of nanoparticles on bilirubin level *in vivo*. Assay results of total and direct bilirubin concentration in the blood specimens of mice in Phase I (**A & B**) and Phase II (**C & D**). In Phase II study, single dose of nanoparticles was able to restore normal serum bilirubin level in less than 6 h from hyperbilirubinemia condition. Data are plotted as mean ± standard error of the mean.

decreased the percentages of AST, ALP and GGT by 53.93, 13.01 and 10.82%, respectively, compared with the CCl₄-treated group. The liver function parameters for the NP control group (group III) remain almost same as the vehicle control group indicating no toxicity of the NPs on liver. Furthermore, no significant change in serum protein concentration indicates the nontoxic effect of NP which is in agreement with the in vitro study involving HSA, described in the earlier section. The statistical analysis of the all liver biochemical-parameters (except BR) is represented in Table 2. The hematological status of animals of group I, group II, group III and group IV has been assessed and are given in the Table 3. It is evident that there is insignificant change in the hematological parameters across the four groups. The result clearly shows that the presence of NPs does not affect hematological system.

The histopathological observations also support the results obtained from serum biochemical assays (Figure 5). The liver lobules of the vehicle control animals (group I) showed a classical structure with hepatocyte plates directed from the portal triads toward the central vein, where they freely anastomose. The liver sinusoids are normally irregularly dilated and spaces of Disse can be seen. The liver sections of this group showed normal hepatic cells, that is, with wellpreserved cytoplasm and prominent nucleus, nucleolus (Figure 5i). The livers of CCl₄-intoxicated mice (group II, Figure 5ii) revealed moderate to severe hepatocellular vacuolization, hepatic necrosis and swelling, bile-duct hyperplasia and increasing cellular mitosis as well as dilation of Disse spaces with focal disruption of the sinusoidal endothelium, inflammatory infiltrations of the portal triads (Figure 5ii) and distortion of the central venules. Mononuclear cell infiltration, hemor-

Table 2. Effect of nanoparticles on ALT, AST, ALP, GGT and total protein in CCl ₄ -intoxicated mice.							
Group	Design of treatment	AST (IU/I)	ALT (IU/I)	ALP (IU/I)	GGT (IU/I)	Total protein (gm/dl)	
I	Sham control	81.43 ± 7.31 ⁺	19.67 ± 1.67 ^{‡,§}	$26.30 \pm 2.63^{\dagger}$	$2.61 \pm 0.30^{\pm,\$}$	6.28 ± 0.54	
II	CCl₄ control	209.78 ± 22.77 ^{‡,§}	58.58 ± 3.28 ^{‡,§}	$54.38 \pm 5.48^{+,\$}$	$5.36 \pm 0.66^{+}$	5.78 ± 0.40	
III	NP control	$79.45 \pm 8.32^{+}$	17.95 ± 1.92 ^{‡,§}	24.58 ± 3.14 ⁺	$2.60 \pm 0.42^{\pm,\$}$	6.12 ± 0.64	
IV	CCl ₄ + NPs	96.65 ± 7.72 ⁺	59.58 ± 3.81 ^{‡,§}	47.30 ± 5.88 ^{+,‡}	$4.78 \pm 0.53^{\ddagger}$	5.88 ± 0.44	
V	CCl_4 + citrate	66.98 ± 10.19 ⁺	42.72 ± 3.28 ^{†,‡,§}	$40.23 \pm 4.41^{+,\pm}$	$4.78 \pm 0.48^{\ddagger}$	5.78 ± 0.44	
VI	CCl ₄ + silymarin	$86.43 \pm 9.44^{\dagger}$	26.79 ± 2.89 ^{†,‡}	$45.98 \pm 4.69^{+,\pm}$	$4.61 \pm 0.47^{t,\pm}$	5.78 ± 0.42	

Data are expressed as mean \pm standard deviation (n = 8)

One-way analysis of variance Tukey post hoc:

[†]p < 0.05 compared with CCl4.CCl₄. [‡]p < 0.05 compared with vehicle control

p < 0.05 compared with vehicle compared silver silver in the silver i

Dosage:- Olive oil: 0.5 ml/kg bodyweight; CCl₄ + olive oil sol.: 1 ml/kg bodyweight; NPs: 500 mg/kg bodyweight; silymarin: 100 mg/kg

bodyweight; citrate (1:1 solution in water): 100 µl/mice

ALP: Alkaline phosphatase; ALT: Alanine aminotransferase; AST: Aspartate aminotransferase; GGT: γ-glutamyltranferase; NP: Nanoparticle.

rhage, fatty degeneration and formation of regenerative nodules were also observed in CCl₄-treated group indicating that CCl₄ induced severe damage to the hepatic cells. The animals treated with NPs (group IV) and silymarin (group VI) (Figure 5iv & vi, respectively) revealed slight to mild hepatocellular vacuolation and better preservation of the normal liver architecture, with moderate hepatocyte plate disorganizations and smaller dilations of Disse spaces. These treated animals displayed rare periportal inflammatory infiltrate in the liver lobules. The animals treated with only NPs (group III) showed normal liver architecture comparable with the vehicle control group. This indicates the nontoxic effect of citrate-capped Mn₂O₄ on hepatocytes which are in good correlation with the liver function tests. It has been reported that CCl₄ causes necrosis, [42,47-50] fibrosis, [47,51-53] mononuclear cell infiltration, [49,53] steatosis and foamy degeneration of hepatocytes, increase in mitotic activity [54] and cirrhosis [42,49] in liver. Therefore, our histopathological findings in the liver due to CCl₄ administration are in agreement with previous studies. However, treatment with NPs significantly decreased these hepatotoxicity characteristics in mouse liver, suggesting that NPs provided protection against CCl₄-induced liver injury. So, according to the microscopic examinations, severe liver damage induced by CCl₄ was remarkably reduced by the administration of the NPs, which was in good correlation with the results of the liver-functional parameters of the serum.

Conclusion

We have synthesized the NPs with the ultrahigh efficacy toward degradation of BR by a simple and green solution-based technique. We further demonstrated

Table 3. Hematological parameters as studied across the group I (sham control), group II (CCI_a control), group III (nanoparticle control), group IV (CCI_a + nanoparticle).

Hematological parameters		oups		
	I	II	III	IV
Hb (gm/dl)	13.57 ± 1.04	13.95 ± 0.89	13.84 ± 0.92	13.90 ± 0.85
RBC (×10º/µl)	6.48 ± 0.56	6.40 ± 0.52	6.45 ± 0.48	6.50 ± 0.58
Rt (%)	1.25 ± 0.19	1.18 ± 0.21	1.22 ± 0.32	1.22 ± 0.23
HCT (%)	40.50 ± 3.42	41.12 ± 3.09	39.85 ± 3.16	40.63 ± 2.13
MCV (fl)	62.72 ± 6.06	64.67 ± 7.81	62.42 ± 5.54	63.02 ± 7.32
МСН (рд)	21.02 ± 2.05	21.94 ± 2.46	21.36 ± 2.48	21.52 ± 2.21
MCHC (gm/dl)	40.50 ± 0.61	33.96 ± 0.85	38.54 ± 3.12	34.23 ± 1.68
Platelets (×10 ³ /µl)	6.53 ± 0.56	6.50 ± 0.55	6.55 ± 0.61	6.42 ± 0.58
WBC (×10⁵/µl)	6.38 ± 0.58	6.65 ± 1.05	6.36 ± 0.95	6.78 ± 0.85

Data are expressed as mean \pm standard deviation (n = 8).

Hb: Hemoglobin; HCT: Hematocrit; MCH: Mean corpuscular hemoglobin; MCHC: Mean corpuscular hemoglobin concentration; MCV: Mean corpuscular volume; RBC: Red blood corpuscle; Rt: Reticulocyte; WBC: White blood corpuscle.

that these Mn-based NPs are safe, biocompatible and effective-targeted probes for hyperbilirubinemia in mice model based on the *in vitro* and *in vivo* assessments, especially without any toxicological implications. The study to compare the efficacy of the NPs with that of a commercially available drug silymarin shows that in hyperbilirubinemia mice, administration of the NPs brings the BR level to normal limits in 6 h whereas silymarin needs several days to control hyperbilirubinemia in mice. The overall therapeutic procedure is pictorially presented in Figure 6. These findings break through the bottleneck in therapeutic procedures for the treatment of hyperbilirubinemia by fast and direct degradation of BR and pave the way for the practical clinical therapy of Mn-based NPs as safe nanomedicines.

Future perspective

In the years to come, advances in engineering nanomaterials with exquisite size and shape control will expand their use in biomedical applications and open the door of personalized medicine. Our study has shown the potential use of Mn-based NPs directly as therapeutic agent against hyperbilirubinemia. The targeted *in vivo* BR degradation ability of surface-functionalized

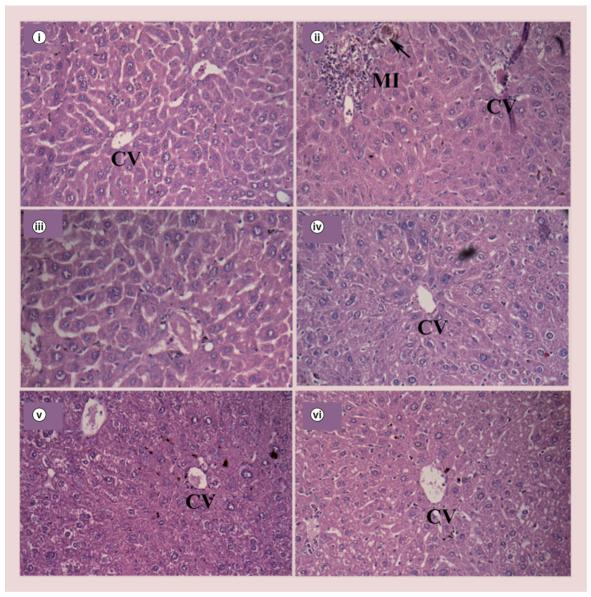


Figure 5. Effect of nanoparticles on hepatic morphological analysis in CCl_4-intoxicated mice. Livers were sectioned and stained with hematoxylin eosin by standard techniques (taken under 100× magnification). Arrows indicate hemorrhage. (I) Sham control, (II) CCl_4 control, (III) nanoparticle control, (IV) nanoparticle treated, (V) citrate control and (VI) Silymarin.

CV: Central vein; MI: Mononuclear cell Infiltration.

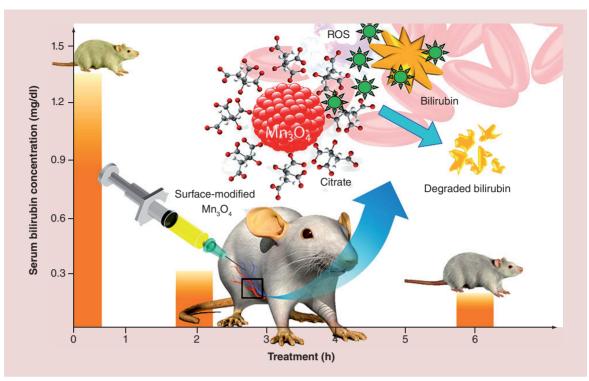


Figure 6. Schematic representation of the nanotherapy using surface-modified Mn₃O₄ nanoparticles. The efficacy of the nanoparticles in reducing serum bilirubin in mice in hours without any side-effect is also represented (see text).

ROS: Reactive oxygen species.

 $\rm Mn_3O_4$ NPs may elucidate a new approach in the symptomatic treatment of hyperbilirubinemia and associated disorders compared with the use of conventional hepa-

toprotective drugs. A detailed toxicological assessment, pharmacokinetic study and experimentation on more robust genetically modified animal models of hyper-

Executive summary

Background

- Hyperbilirubinemia is one of the most common diseases in tropical countries causing several neurologic damages for neonates as well as adults and if untreated may lead to morbidity and even mortality.
- Current therapeutics for hyperbilirubinemia are poorly effective, nontargeted and time consuming. Also there is no medicine available which can degrade bilirubin directly in *in vivo* condition.

Outcomes of the study

 In this study, the authors demonstrate that intraperitoneally injected citrate-functionalized Mn₃O₄ nanoparticles (NPs) degrade bilirubin very fast and in a specific, nontoxic way compared with conventional drugs in a mouse model.

Methods

- Synthesis and characterization of citrate-functionalized Mn₃O₄ NPs have been done.
- The *in vitro* catalytic activity of NPs in degradation of bilirubin through reactive oxygen species mechanism has been established.
- In vivo preclinical studies of Mn₃O₄ NPs as a therapeutic agent against hyperbilirubinemia were done using mice as a model organism.
- The efficiency of Mn₃O₄ NPs in treatment of hyperbilirubinemia in mice is ensured by biochemical tests and histopathological studies.

Conclusion & future perspective

- Our results confirmed that Mn-based NPs are nontoxic, biocompatible and effective targeted probes against hyperbilirubinemia.
- This study may open a new door for faster, safer and efficient therapeutic treatment of hyperbilirubinemia, jaundice and associated diseases.
- This approach may be applied for future nanomedicine applications.

bilirubinemia (e.g., UGT1 A1 deficient mice or Gunn rat model of neonatal jaundice) will help to verify the potential of this nanoparticle in preclinical studies more strongly and lead the way to clinical trials. Finally, it is concluded, in view of good safety margin and faster action with the proposed nanomedicine (NPs), that the medicine could become the symptomatic treatment of choice in hyperbilirubinemia in general.

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Financial & competing interests disclosure

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Ethical conduct of research

The authors state that they have obtained appropriate institutional review board approval or have followed the principles outlined in the Declaration of Helsinki for all human or animal experimental investigations. In addition, for investigations involving human subjects, informed consent has been obtained from the participants involved.

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